Palladium- and Platinum-Catalyzed Silaboration of Methylenecyclopropanes through Selective Proximal or Distal C-C Bond Cleavage

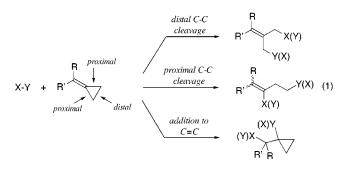
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Methylenecyclopropanes (MCPs) are highly strained but readily accessible molecules that have served as useful building blocks in organic synthesis.¹ Especially, increasing attention has been paid to the transition-metal catalyzed reactions of MCPs, which have been employed for construction of complex organic molecules.² The use of the transition-metal catalysts for the activation of MCPs may involve σ -bond metathesis reactions with concomitant C–X and C–Y bond formations in the presence of appropriate reactants X–Y such as R₃Si–CN,³ R₃Si–H,⁴ R₃Sn– H,⁵ RO–H,⁶ R₂N–H,⁷ and R₂B–BR₂.^{8,9}

An attractive but often troublesome feature of MCPs is their multiform reactivities that may lead to formation of a variety of products (eq 1). Besides the three σ -bonds (two proximal and



one distal bonds) in the cyclopropane ring to be cleaved, the C= C bond may undergo the addition reaction with X-Y in the presence of the transition-metal catalysts.^{3,5,8} Furthermore, for the reactions with unsymmetrical X-Y reagents, the regiochemistry, with which X and Y are introduced to the inequivalent carbon atoms, may make the reaction complicated. Although some σ -bond metathesis reactions of MCPs were found to proceed in a selective fashion to afford one of those possible products, selective synthesis of different products from common reactants by choice of catalyst has never been reported.^{2,10} Such synthetic transformations may be highly important from the viewpoint of transition-metal catalysis as well as organic synthesis.

(1) Synthesis of MCPs: Brandi, A.; Goti, A. Chem. Rev. **1998**, 98, 589. (2) Reviews: (a) Binger, P.; Büch, H. M. Top. Curr. Chem. **1987**, 135,

77. (b) Ohta, T.; Takaya, H. In *Comprehensive Organic Synthesis*; Trost, B. M., Ed.; Pergamon: Oxford, 1991; Vol. 5, p 1185.

(3) Chatani, N.; Takeyasu, T.; Hanafusa, T. Tetrahedron Lett. 1988, 29, 3979.

(4) Bessmertnykh, A. G.; Blinov, K. A.; Grishin, Yu. K.; Donskaya, N. A.; Tveritinova, E. V.; Yur'eva, N. M.; Beletskaya, I. P. J. Org. Chem. 1997, 62, 6069.

- (5) Lautens, M.; Meyer, C.; Lorenz, A. J. Am. Chem. Soc. 1996, 118, 10676.
 (6) Camacho, D. H.; Nakamura, I.; Saito, S.; Yamamoto, Y. Angew. Chem., Int. Ed. 1999, 38, 3365.
- (7) Nakamura, I.; Itagaki, H.; Yamamoto, Y. J. Org. Chem. 1998, 63, 6458.
 (8) Ishiyama, T.; Momota, S.; Miyaura, N. Synlett 1999, 1790.

(9) For the related σ -bond metathesis reactions of MCPs with malonates

and derivatives, see: Tsukada, N.; Shibuya, A.; Nakamura, I.; Yamamoto, Y. J. Am. Chem. Soc. **1997**, 119, 8123.

(10) Such examples have been known for the cycloaddition reactions: (a) Bapuji, S. A.; Motherwell, W. B.; Shipman, M. *Tetrahedron Lett.* **1989**, *30*, 7107. Compare the following reports: (b) Noyori, R.; Kumagai, Y.; Umeda, I.; Takaya, H. J. Am. Chem. Soc. **1972**, *94*, 4018. (c) Binger, P.; Schuchardt, U. Angew. Chem. **1977**, *89*, 254.

We have developed transition-metal catalyzed reactions of silylboranes with unsaturated organic compounds for selective introduction of silicon- and boron-functionalities to the carbon- carbon multiple bonds, leading to the regio- and stereoselective synthesis of new silylated organoboron compounds.^{11,12} Herein, we disclose palladium- and platinum-catalyzed reactions of MCPs with silylborane **1**, in which appropriate choice of metals and ligands selectively promoted the cleavage of the proximal or the distal C–C bond of MCPs, giving the corresponding silaboration products regio- and stereoselectively.

We initially examined the reaction of ethyl cyclopropylideneacetate (**2a**) with **1** in the presence of a palladium/*t*-OcNC catalyst (*t*-OcNC: 1,1,3,3-tetramethylbutyl isocyanide), which was reported as an effective catalyst for silaborations of carbon–carbon multiple bonds with **1**.^{12b,c} The reaction was completed within 4 h at 110 °C in toluene to afford α , β -unsaturated ester **3a** in 89% yield as a 85:15 mixture of *Z* and *E* isomers (eq 2). The product

$$PhMe_{2}Si-B \xrightarrow{O}_{1}$$

$$1$$

$$CO_{2}Et$$

$$2a \xrightarrow{CO_{2}Et}_{catalyst}$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$CO_{2}Et$$

$$SiMe_{2}Ph$$

$$(2)$$

$$3a$$

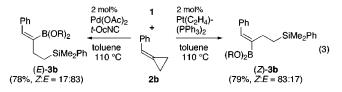
$$3a$$

$$Catalyst: Pd(OAc)_{2}/t \cdot OcNC (89\%, Z:E = 85:15)$$

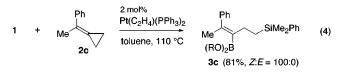
$$: Pt(C_{2}H_{4})(PPh_{3})_{2} (74\%, Z:E = 74:26)$$

possessed a vinylic boryl group as well as a homoallylic silyl group in the molecule, indicating the proximal C-C/Si-B metathesis proceeded regioselectively. The exclusive proximal C-C bond cleavage with an introduction of the boryl group to the sp² carbon was also achieved with $Pt(CH_2=CH_2)(PPh_3)_2$.

Of interest is that the stereoselective silaborative cleavage at the proximal C–C bond of **2b** either *cis* or *trans* to the phenyl group depended upon the catalyst employed. While the platinum catalyst gave (Z)-**3b** selectively via a cleavage of the proximal C–C bond *trans* to the phenyl group, the palladium catalyst afforded the corresponding *E* product predominantly via the proximal *cis*-C–C bond cleavage (eq 3).



Higher selectivity for the Pt-catalyzed cleavage at the less sterically hindered proximal C-C bond was observed in the reaction of 2c, giving alkenylborane (Z)-3c in high yield as a sole product (eq 4). With this substrate, however, use of the



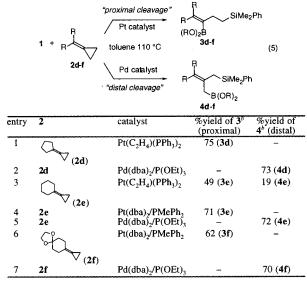
palladium-isonitrile catalyst resulted in low conversion with formation of a complex mixture of products.

Similarly, **2d** reacted with **1** in the presence of the platinum catalyst, resulting in the selective formation of alkenylborane **3d** via the proximal C-C bond cleavage (eq 5; Table 1, entry 1). Although the palladium-isonitrile catalyst failed to achieve high

⁽¹¹⁾ Suginome, M.; Ito, Y. Chem. Rev. 2000, 100, 3221.

⁽¹²⁾ For recent examples, see: (a) Suginome, M.; Matsuda, T.; Yoshimoto, T.; Ito, Y. Org. Lett. 1999, 1, 1567. (b) Suginome, M.; Ohmori, Y.; Ito, Y. Synlett 1999, 1567. (c) Suginome, M.; Matsuda, T.; Nakamura, H.; Ito, Y. Tetrahedron 1999, 55, 8787.

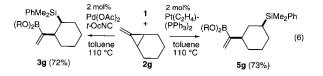
Table 1. Reactions of Cycloalkylidenecyclopropanes 2d-f with Silylborane 1 in the Presence of Palladium and Platinum Catalysts^{*a*}



^{*a*} A mixture of **1**, **2** (1.2–2.0 equiv), and the catalyst [metal/P = 1/2, 2 mol % (entries 1–3) or 5 mol % (entries 4–7) of the metals] was heated in toluene at 110 °C. ^{*b*} Isolated yield.

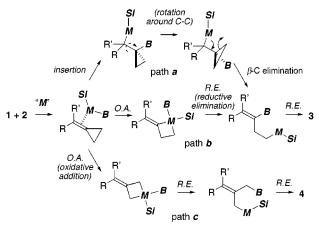
conversion, we found that a catalyst prepared from $Pd(dba)_2$ (dba: dibenzylideneacetone) with $P(OEt)_3$ (2 equiv to Pd) promoted the reaction of **2d** with **1** effectively (entry 2). To our surprise, however, the silaboration product **4d** was derived regioselectively from the distal C–C bond cleavage. The exclusive distal cleavage was also observed in the reactions of related cycloalkylidenecyclopropanes **2e** and **2f** in the presence of the palladium-phosphite catalyst system (entries 5 and 7). It should be noted that a reaction of **2e** with **1** in the presence of $Pt(C_2H_4)$ -(PPh₃)₂ resulted in the formation of a mixture of proximal and distal bond cleavage products with the former predominating (entry 3). It was found that $Pt(dba)_2/PMePh_2$ catalyzed selectively the proximal C–C cleavage, giving **3e** in good yield (entry 4).¹³ The catalyst system was successfully applied to the selective synthesis of **3f** without formation of **4f** (entry 6).

Finally, we refer to the reaction of 2g, which has substituents on the cyclopropane ring (eq 6). The palladium-isonitrile catalyst



induced the silaborative proximal C–C bond cleavage effectively to give 1-organosilyl-2-(α -borylvinyl)cyclohexane **3g** as a single *cis* isomer. In contrast, a platinum catalyst provided *cis*-1,3-disubstituted cyclohexane **5g** stereoselectively in good yield, which may also be derived via the proximal C–C bond cleavage.

Scheme 1. Possible Mechanisms (Paths a-c) for the Reactions of MCPs (2) with 1^a



^{*a*} Si, B, and M stand for SiMe₂Ph, B(OR)₂, and ML*n*, where M = Pd or Pt.

For the closely related reactions of MCPs with hydrosilanes (Rh cat.),⁴ tributylstannane (Pd cat.),⁵ and diboron (Pt cat.), which proceed through the proximal bond cleavage,⁸ a mechanism involving migratory insertion of the C=C bond of MCP and subsequent β -carbon elimination (e.g., path *a* in Scheme 1) has been proposed for the formation of intermediary homoallylic transition-metal complexes. The unexpected formation of 5g may arise from β -hydride elimination from the corresponding homoallylic platinum intermediate and subsequent re-addition of the H-Pt with the opposite direction.¹⁴ Concerning the mechanism involving path a, the E/Z selectivity in the reactions of unsymmetrically substituted MCPs such as 2b and 2c is determined by the direction of σ -bond rotation prior to the β -C elimination step. The rotation should occur so as to avoid steric repulsion, favoring the formation of Z isomers regardless of the nature of the metal. However, this mechanism can hardly explain the observed higher Z-selectivity for the Pt-catalyzed reaction of 2c than that for 2b, since 2c has less pronounced difference in steric bulkiness between R and R' than 2b. We may suggest another mechanistic possibility, which involves oxidative addition of the proximal C-C bond to the transition-metal (path b). The selectivity of the C-C bond cleavage can be affected by the electronic nature of R and R'. The mechanism may be operative predominantly, when the migratory insertion step for path a is unfavorable due to the substituent(s) at the C = C bond as well as the nature of the metals and ligands. The direct oxidative addition mechanism may also be involved in the distal bond cleavage (path c).¹⁵

The present reactions of MCPs with **1** may be utilizable for preparation of a variety of synthetically useful silyl-substituted alkenylboranes and allylboranes, which are otherwise inaccessible.

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Supporting Information Available: Experimental procedures including those of transformation for characterization and spectral data for the new compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹³⁾ Use of $Pt(dba)_2$ with no additional ligands resulted in the formation of a complex mixture. See ref 8.

⁽¹⁴⁾ Suginome, M.; Nakamura, H.; Ito, Y. Angew. Chem., Int. Ed. Engl. 1997, 36, 2516.

⁽¹⁵⁾ For theoretical studies on transition-metal catalyzed reactions of MCPs with unsaturated organic molecules, see: Suzuki, T.; Fujimoto, H. *Inorg. Chem.* **2000**, *39*, 1113 and references therein.